



CHEMSTREAM

SUSTAINABLE CHEMISTRY

GLASSPrint2019
CONFERENCE

Adhesion of UV-curable inks and coatings on glass

Dr. Marin Steenackers
ChemStream



ChemStream: The Independent Chemical R&D Company

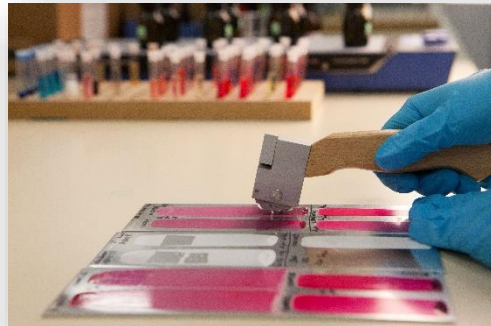
Translating customized requirements into chemical formulations with dedicated functionality, from **design to prototyping and implementation**

- Core activities:
 - Innovative contract research
 - Customized product development
 - Design and synthesis of functionalized (bio-based) polymers (dispersants, emulsifiers, surfactants...)
- Main deliverables:

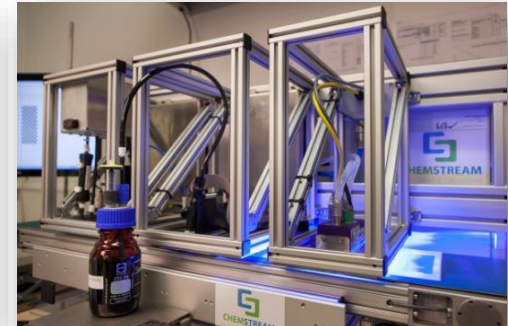
- **Nano dispersions**



- **Coatings**



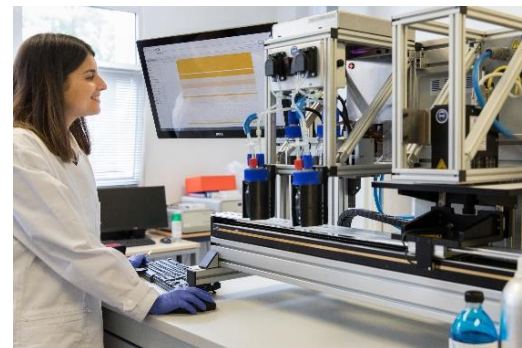
- **Inkjet inks**





ChemStream: The Independent Chemical R&D Company

- **Founded in April 2010**
- **Staff profile (14 FTE, 11 PhDs)**
 - Chemistry (12)
 - Material Science (1)
 - Bio Engineer (1)
- **Located near Antwerp – Belgium**
- **Lab-facilities (550 m²)**
 - Organic Synthesis
 - Chemical Formulation
 - Characterization
- **Prototype production facility**
 - Coatings: 250 L batches
 - Inkjet inks: 25 L batches





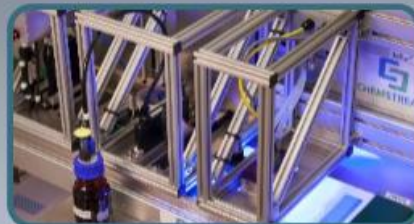
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Expertises



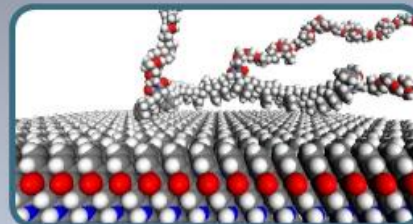
Organic Synthesis

- * Crystal, colorant and dispersant design
- * Photochemistry
- * Interfacial chemistry, wetting and adhesion
- * Superabsorbing polymers
- * Flow chemistry



Technology

- * Dispersion technology
- * Coating, printing, jetting (Modular printing unit MPU)
- * Radiation curing (UV, UV-LED, e-Beam)
- * Atmospheric plasma



Methodology

- * Molecular modeling
- * Design of Experiments (DoE)
- * Smart throughput screening
- * Hansen solubility parameters (HSP)



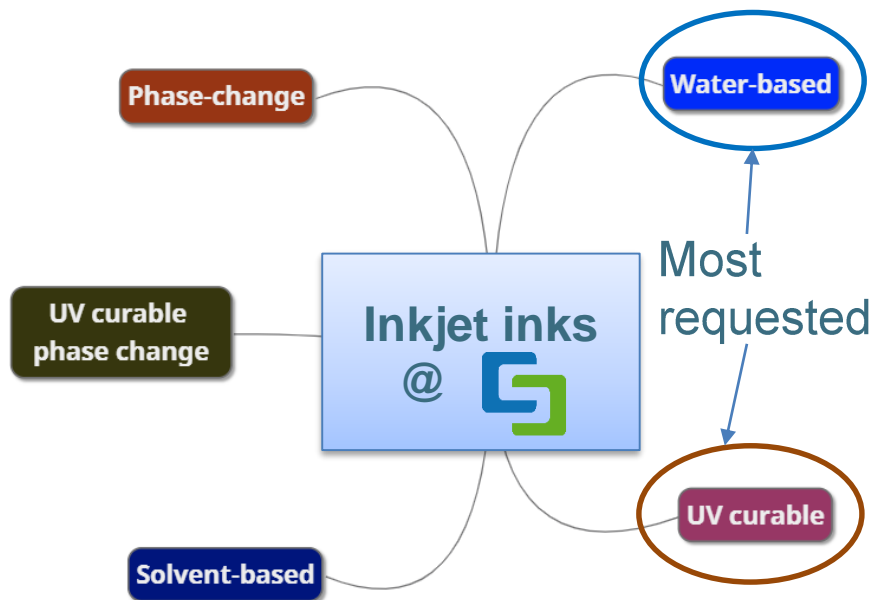
Analytical and physical chemical tools

- * UVVIS, FTIR, GCMS, LCMS, GPC
- * Particle size distribution (PSD)
- * Contact angle, surface tension, viscosity



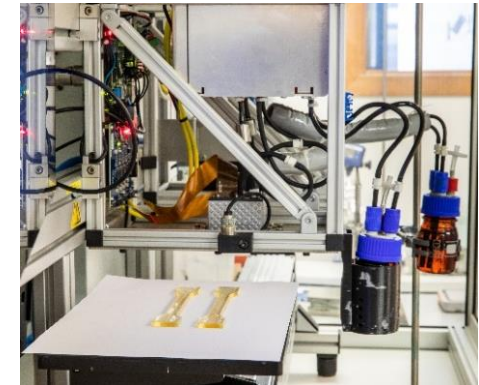
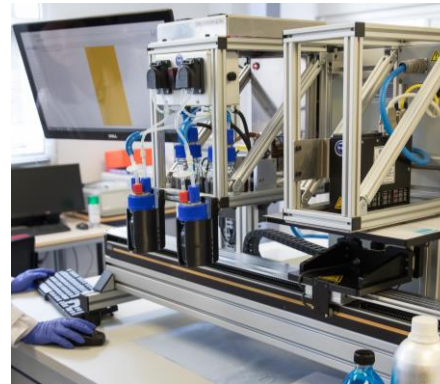
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Inkjet @ ChemStream



Modular Printing Units

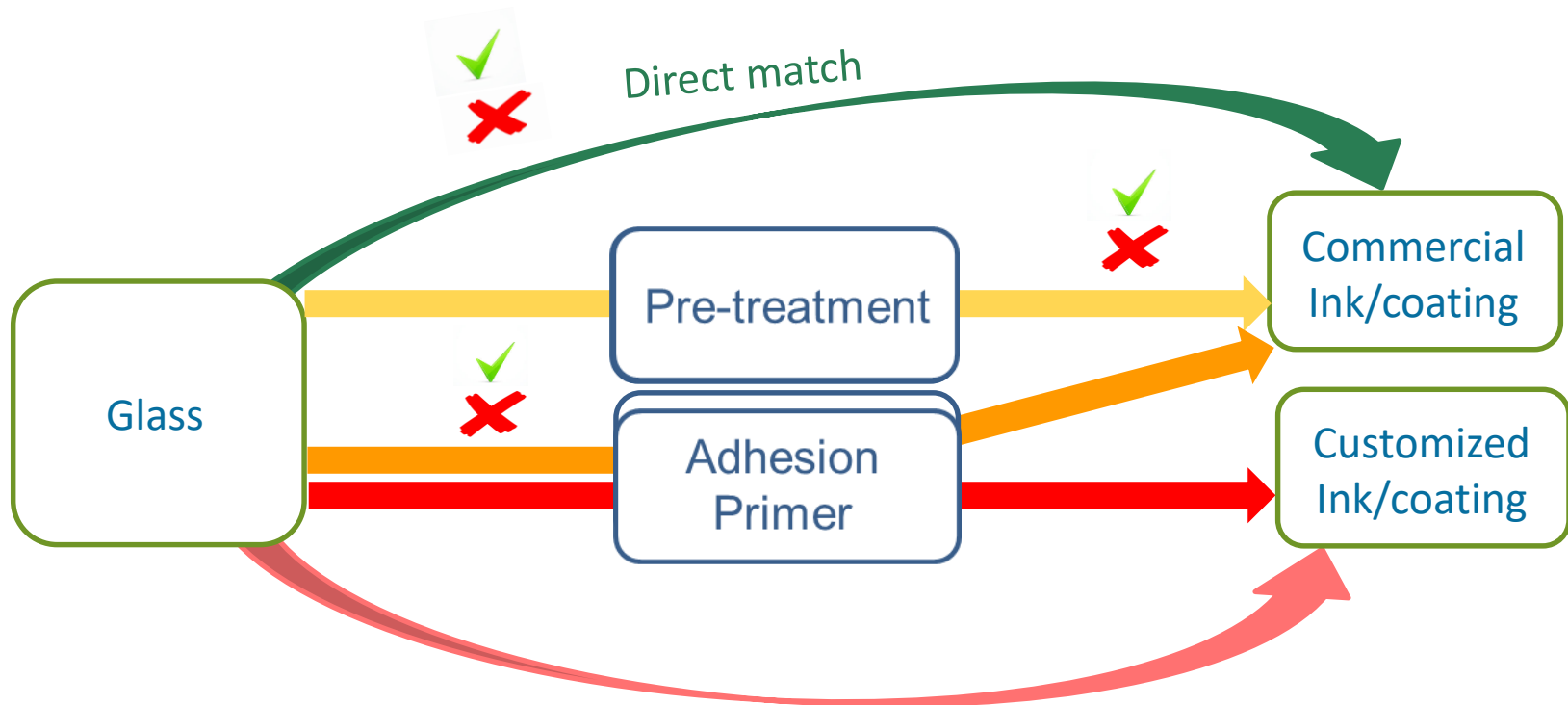
- Mimic of an in-line printing process
- Fast iterations of ink prototypes
- Different inkjet printheads
- 3D printing





Adhesion: the challenge

Direct to object adhesion



Technology: UV curable inks, coatings, ...



Adhesion: the challenge

Direct to object reversible adhesion

→ Tunable Deinking }
→ Tunable Stripping } E.g. in acid/base, @ high T, etc.

Examples:

Printing on returnable bottles



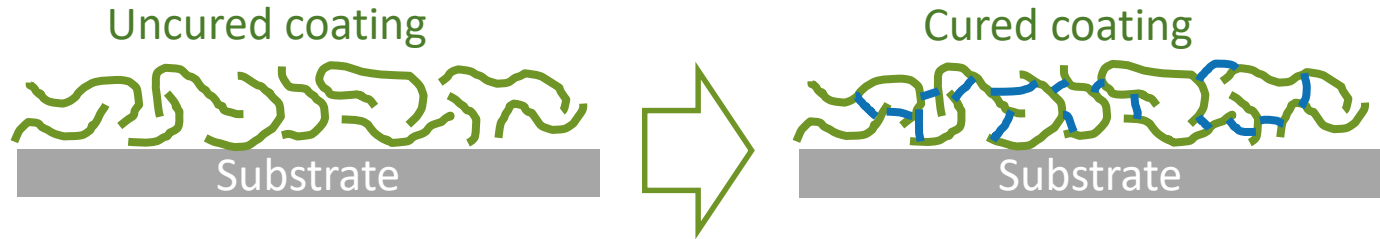
Temporary mask –
frosted glass



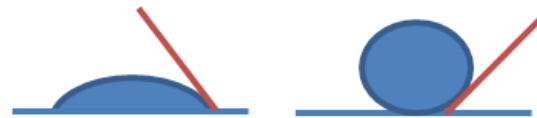


Factors influencing adhesion

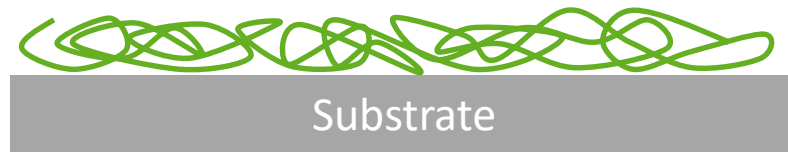
➤ Shrinkage



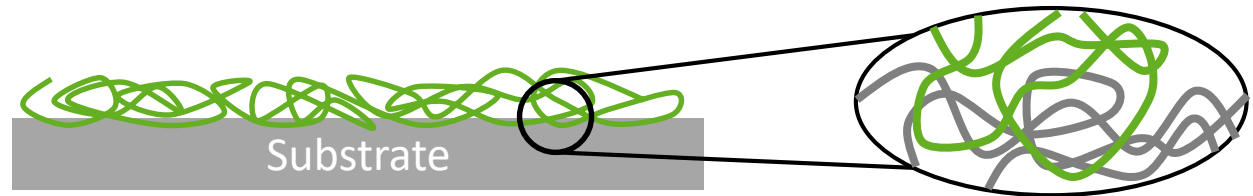
➤ Wetting



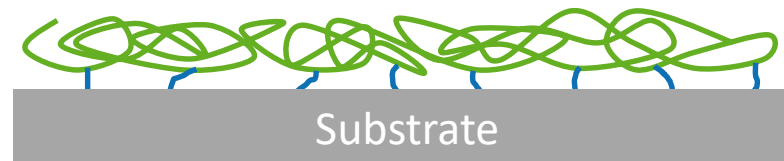
➤ Non-covalent interactions



➤ Entanglement



➤ Covalent bonding



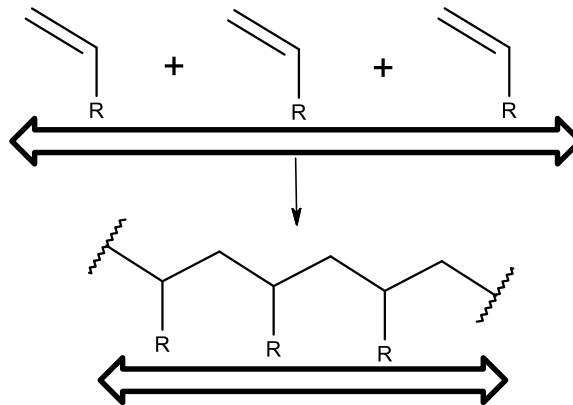


Adhesion: Shrinkage

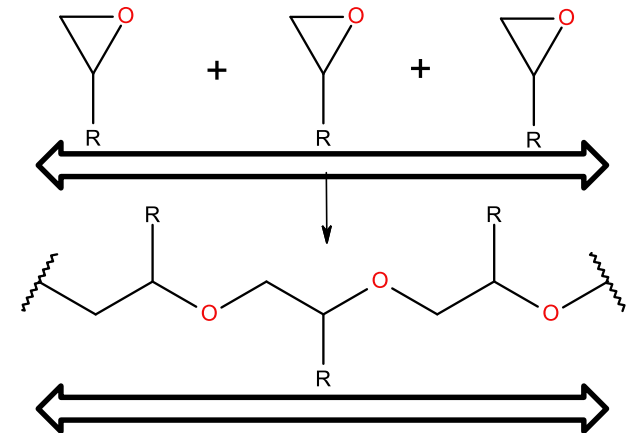
Property	Radical Polymerization	Cationic Polymerization
Oxygen Inhibition	Yes	No
Inhibition by humidity	No	Yes
Cure speed	Milliseconds	Seconds
Shrinkage	> 10 %	< 10 %
Adhesion	Good	Very good
Raw Material Cost	Acceptable	High

Shrinkage:

(Meth)acrylates



Epoxides

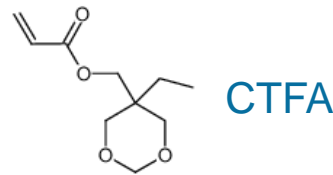
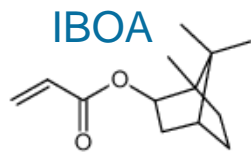




Adhesion: Shrinkage

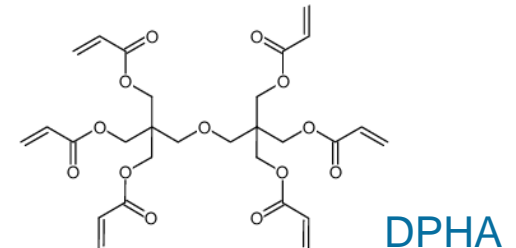
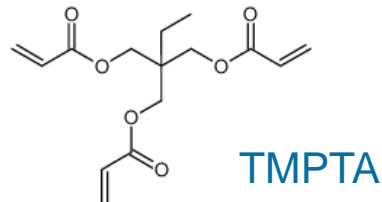
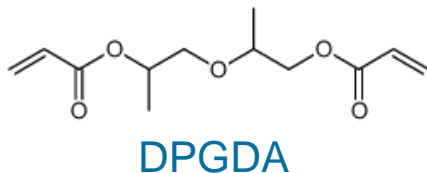
How to minimize shrinkage for (meth)acrylates

- ✓ Use of monofunctional monomers



- ✓ Use of multifunctional oligomers (high viscosity )

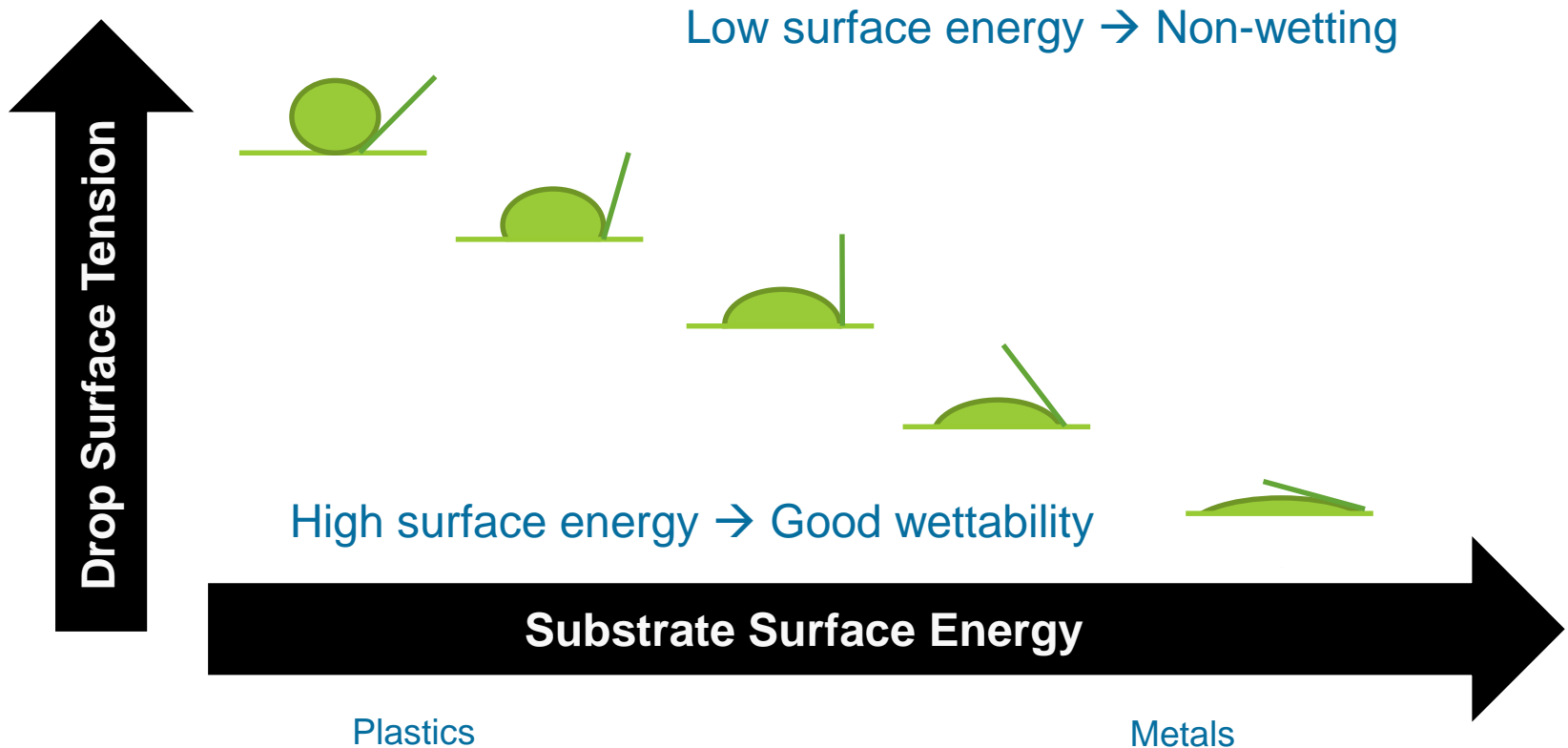
- ✓ Minimize the use of small crosslinkers





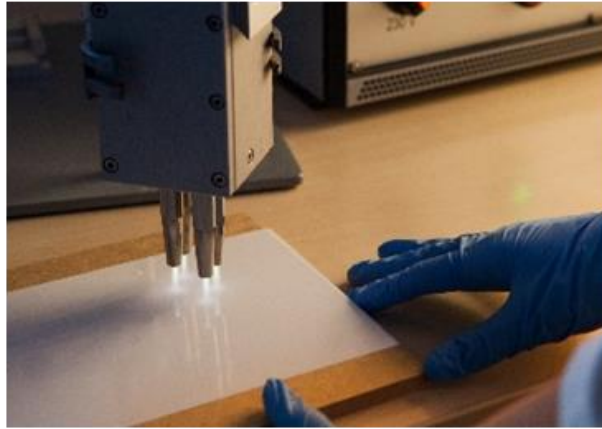
Adhesion: Wetting

Substrates

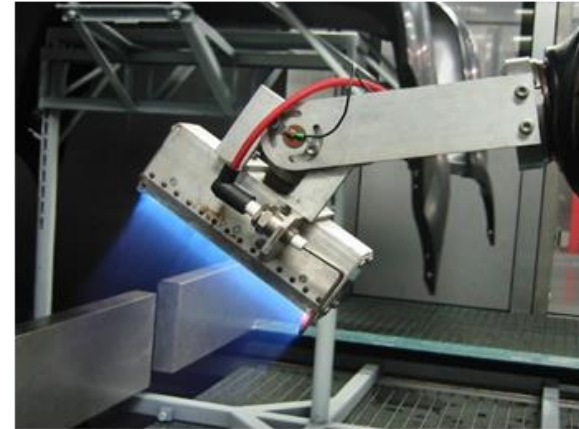




Adhesion: Improve wetting by pre-treatment



Atmospheric plasma



Flaming

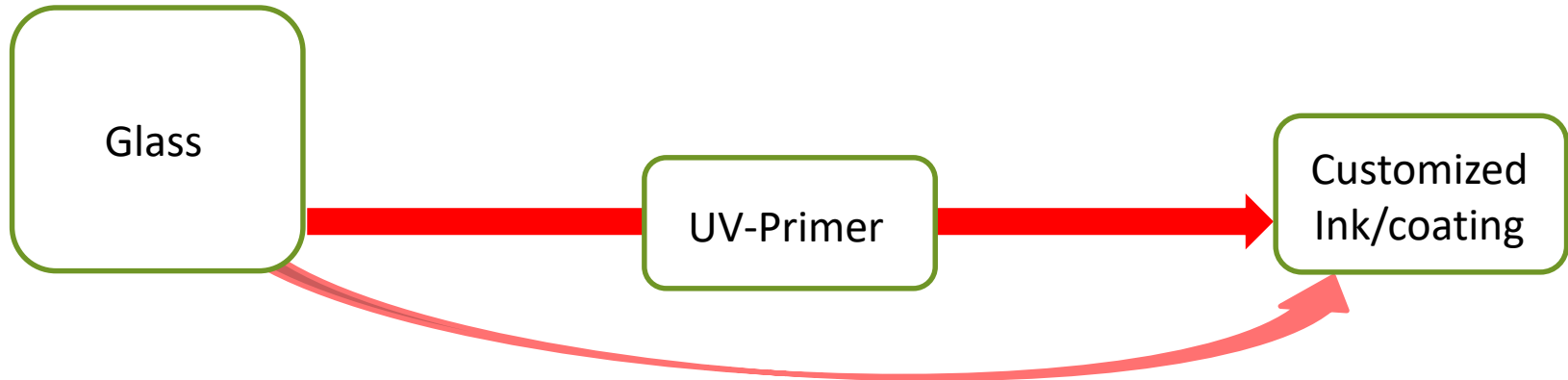
Surface activation

Surface cleaning

Removal of oil, dust and organic coatings (CEC)



Adhesion: Covalent bonding

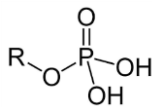


Ink formulation:

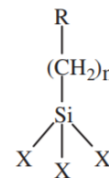
- ✓ Use of monomers to minimize shrinking → better adhesion
- ✓ Use of monomers with strong non-covalent interactions (polar, H-bonding)
- ✓ Use of adhesion promoters needed for a good adhesion (mostly acidic)

Phosphate & Silane best options (mono and multidentate)

Phosphate
functional



Silane
functional





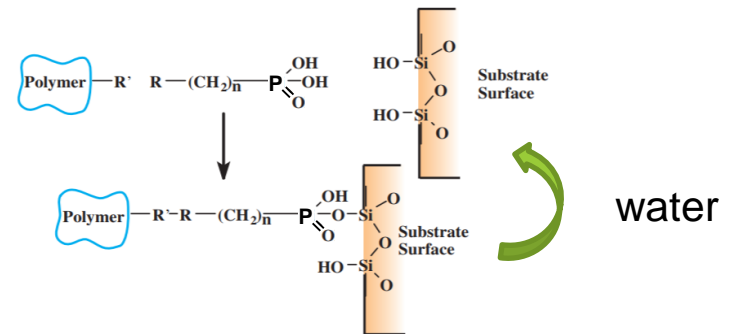
Adhesion: Covalent bonding

Phosphate functional groups

- ✓ Adhesion promoter with strongest adhesion on glass.
- ✓ Drawback → Phosphates desorb in prolonged contact with water.



Printed beer glass → Not suitable option → e.g. dishwasher removes ink



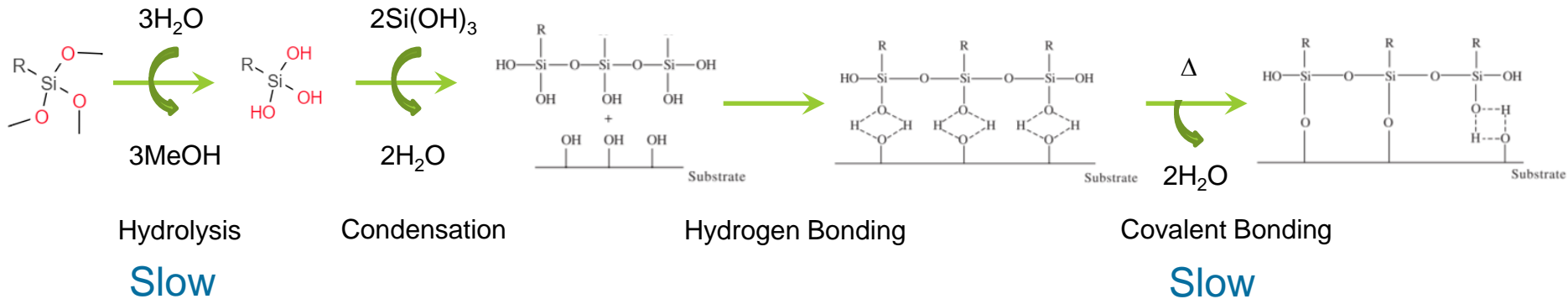
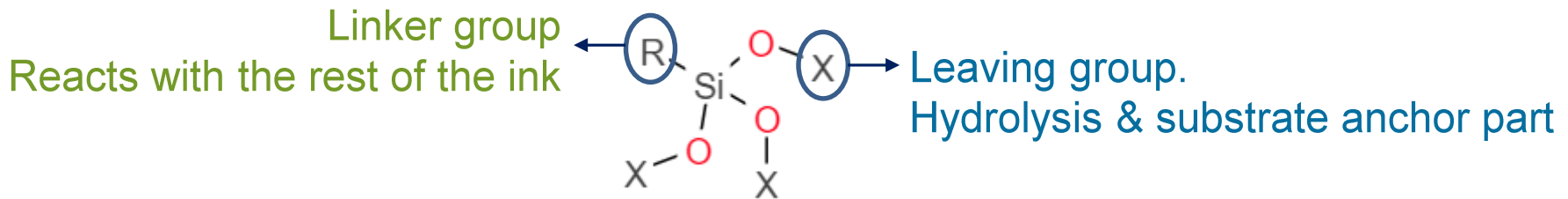
→ Perfect for applications which require a good dry adhesion



Adhesion: Covalent bonding

Silane functional groups

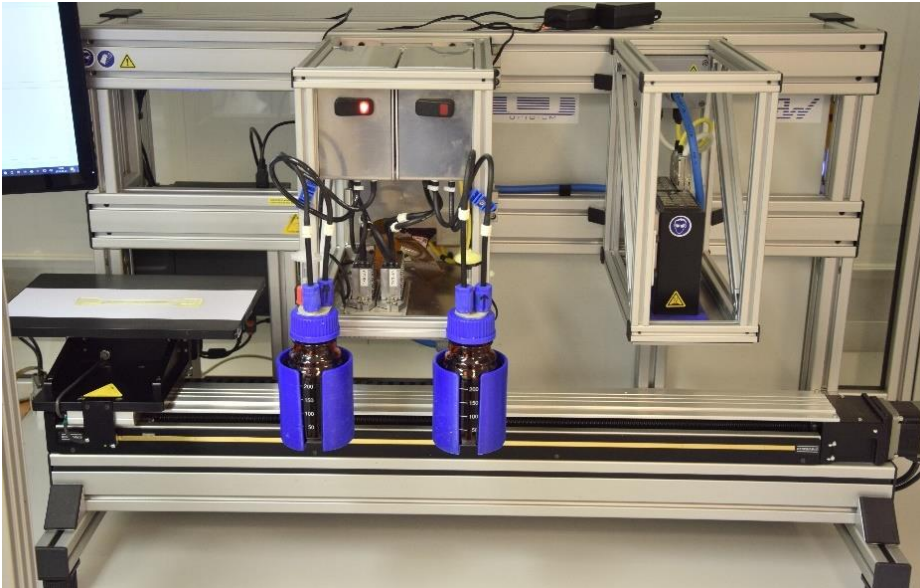
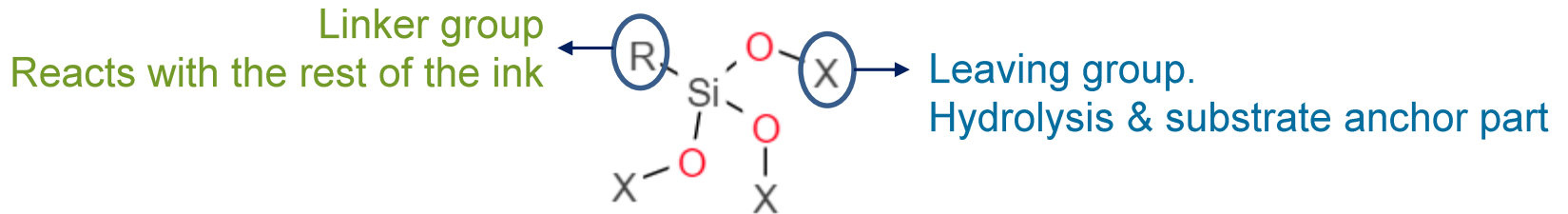
- ✓ Once the covalent bond is formed, excellent & resistant adhesion.
- ✓ Drawback: Kinetics are very slow. Formulation adapted to speed up the hydrolysis and condensation of the silanes (e.g. catalyst)





Adhesion: Covalent bonding

Silane functional groups

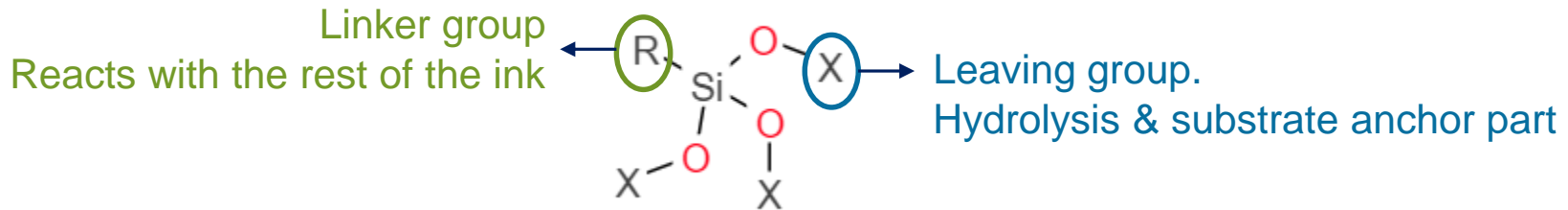


→ Anchoring has to take place during period between jetting and pinning



Adhesion: Covalent bonding

Silane functional groups

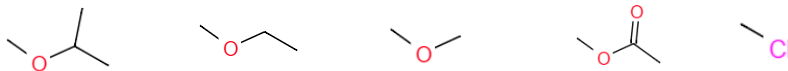


Linker Group → Chosen depending the composition of the ink.
UV curable ink: e.g. (Meth)acrylic, Vinyl, thiol, amine, ...

Leaving group → Trade-off between reactivity & stability

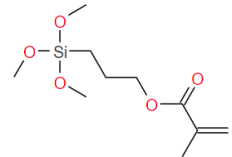
Reactivity (Hydrolysis): $\text{Cl}^- > \text{AcOH} > \text{MeOH} > \text{EtOH} > \text{iPrOH}$

Stability: $\text{iPrOH} > \text{EtOH} > \text{MeOH} > \text{AcOH} > \text{Cl}^-$



Same compromise for catalysed silanes

→ Anchoring reaction is bottle neck towards adhesion



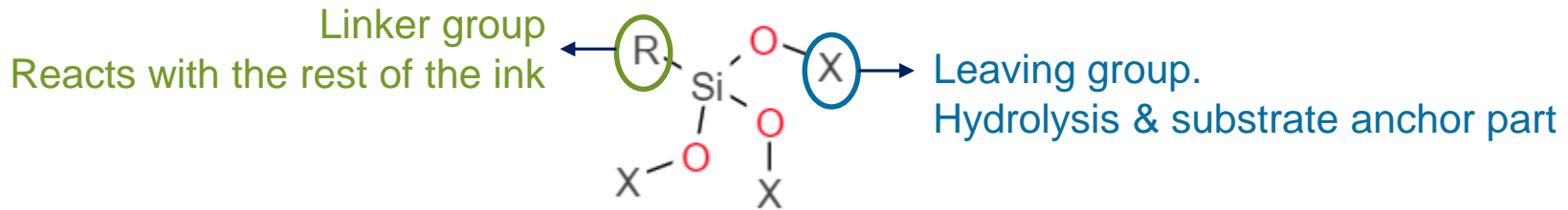
e.g. “MEMO”

- ✓ Price
- ✓ Ink interaction
- ✓ Slow hydrolysis



Adhesion: Covalent bonding

Silane functional groups



ChemStream → synthesis of a new class of silane-based adhesion promoters to adhere directly on glass for UV-curable primers, inks and coatings



Thanks for your attention !!!

You are invited at our booth 23
www.chemstream.be

